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Production of Antioxidant Bioactive Compounds during Mycelium Growth of *Schizophyllum commune* on Different Cereal Media

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Abstract Schizophyllum commune or split gill mushroom is a nutrient-rich natural food ingredient. Even during mycelium growth, S. commune could produce the bioactive compounds including phenolic and flavonoid compounds. The aim of this research was to characterize the production of bioactive compounds during mycelium growth period of *S. commune* cultured onto different eight edible cereal media (sorghum, corn, barley, wheat, oat, jasmine rice, Mun Poo rice and riceberry rice). The antioxidant activities and mycelial biomass were also observed during their growth. The results show that the highest mycelial growth rate was obtained onto barley and jasmine rice, while the lowest one was found in sorghum and corn. The concentration of phenolic compounds increased along with the mycelial growth. The fungal culture on wheat significantly exhibited the highest production of phenolic compounds which was 8.56 ± 1.09 mg GAE/g DW on day 8. The highest flavonoid production of 577.35 \pm 29.93 µg CE/g DW, was remarkably found onto barley from day 6. The degradation of certain flavonoids in cereal materials by S. commune was also observed, particularly in riceberry rice. Antioxidant activity of cultured cereals was depended on the initial activities of materials and was mainly increased by S. commune metabolism. The DPPH• and ABTS⁺ scavenging activities were sharply increased at day 4 which was early stage of log phase of S. commune. At day 8, most of cereal media exhibited high DPPH• activity with a half maximal inhibitory concentration (IC_{50}) range of 3.83-5.80 mg DW/mL, except jasmine rice. Only wheat and oat could significantly give the highest ABTS⁺⁺ scavenging activity which was in an IC₅₀ range of 2.38-3.27 mg DW/mL. The highest FRAP activity with a median effective concentration (EC₅₀) of 2.14 ± 0.23 mg DW/mL, was observed in barley which corresponded with its highest flavonoid content. Antioxidant activities of S. commune culture onto cereal media were correspondent with their phenolic and flavonoid contents. Therefore, this study assured that several antioxidative substances were interestingly produced by using *S. commune* cultured onto the selected edible cereals, which could be possibly developed as a new supplement or active ingredient for pharmaceutical and food industries.

Keywords: Antioxidant activity, Bioactive compound, Cereal, Schizophyllum commune, Split gill mushroom

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INTRODUCTION

Schizophyllum commune or commonly known as split gill mushroom is an edible species of Basidiomycetes. This rubber wood-decaying mushroom is predominantly found in the rainforest of southern Thailand. This mushroom has been used as a medicine for a long time because its fruiting body contains several biologically active compounds, for instance, phenolic compounds, vitamins, carboxylic esters and β -glucan namely schizophyllan. These bioactive compounds could exhibit the promising health benefits such as immune system stimulation (Batbayar et al., 2012), antioxidant activity (Wanna et al., 2018), antitumor activity and suppression of inflammation (Takedatsu et al., 2012). Therefore, this mushroom or its isolated fractions have been widely applied in cosmetic, medical, pharmaceutical, nutraceuticals and food industries. Even, during the *S. commune* mycelial growth period in liquid bacterial media, it could produce the phenolic compounds which showed the radical scavenging activities (Dulay et al., 2016). Phenolic compounds are secondary metabolites, which found in most abundantly in plants, fungi, and bacteria. They play an important role as the defense mechanisms.

The difference in nutrient composition in culture media is an important factor on the mycelial growth and antioxidant synthesis (Hoa et al., 2015). Generally, sorghum or corn is mostly used as a mushroom spawn in Thai mushroom farming because of their cost- and nutrient-effectivenesses for most mushroom fungi growing. Lee et al. (2014) documented that the characteristics of various cereal grains such as size, shape and nutrient content, have an influence on mycelial growth characteristics. Edible cereals in Thai agricultural use, for example, riceberry rice, Mun Poo rice and jasmine rice have been reported in different nutrition compositions (Wangpakapattanawong, 2010). Non-starch polysaccharides are main component in cell wall which consists of cellulose, xylose, arabinose, β -glucan and non-carbohydrate such as phenolic acids and lignin. For example, wheat and corn are rich in arabinoxylan consisting of arabinose and xylose copolymer, whereas barley and oat present a high 1,4 β -glucan content (Knudsen, 2014).

Free radicals are unstable electrons in the atomic orbitals which considered as an important factor affecting to the human health . These radicals could be generated from unhealthy behaviors and pollutions and cause oxidative damage on human cellular DNA and membrane of mitochondria, which consequently leads to the chronic diseases such as endothelium dysfunction, atherosclerosis, diabetes, cardiovascular disease and also the process of aging (Ghebre et al., 2016). Therefore, the body requires antioxidant substances to balance the free radicals and inhibit the oxidative stress. The previous studies reported that the high antioxidant phenolic and flavonoid compounds are found in the mycelium of several edible mushrooms (Vamanu, 2014). The study of Dulay et al. (2016) found that the nutritional condition in substrate had influence on mycelium growth, ability to synthesize the bioactive compounds, and antioxidant properties of S. commune mycelium Hence, this research aimed to exhibit the effect of nutrient composition in edible cereals media on production of mycelial biomass along with antioxidant properties during the mycelium growth period of S. commune, which could be possibly applied and developed in the commercial production process for pharmaceutical and food industries.

MATERIALS AND METHODS

Materials

The starter culture of *S. commune* was obtained from Chai-Yo farm, Suratthani, Thailand. The cereals used as culture media in this study were purchased from different local markets in Bangkok, Thailand, including whole grain sorghum (*Sorghum bicolor*), whole grain yellow dent corn (*Zea mays* var. *indentata*), pearl barley (*Hordeum vulgare*) (Raitip, Thanya Farm, Nonthaburi, Thailand), whole grain wheat (*Triticum aestivum*) (Dr.Green, Bangkok, Thailand), whole grain oat (*Avena sativa*) (Home Fresh Mart, Capmax Trading, Bangkok, Thailand), white jasmine rice (*Oryza sativa*) (J.P. Rice International, Surin, Thailand), partially-polished Mun-Poo rice (Naturezone, Bangkok, Thailand), and whole grain riceberry rice (Rice for Health, Shaiyo Triple A Group, Chonburi, Thailand). Potato dextrose agar (PDA) was purchased from Oxoid (Hampshire, UK). All analytical chemicals were purchased from Sigma Chemical (St Louis, MO, USA).

Cultivation of S. commune on cereal media

The preparation of cereal media, 30 g of each cereal material was immersed in water, until obtaining 50% moisture which separately measured by the gravimetric method with oven drying (AOAC, 2005). Each wet cereal was steamed until it was fully cooked in a range 20-30 min. Each cooked cereal was transferred into a flat bottle and sterilized at 121°C for 15 min. Obtained *S. commune* culture was isolated on PDA in a flat bottle and then incubated at 30°C for 5 days. The isolated fungal colony was cut by using an inoculation loop in triangle shape of 0.5 cm in each side and then transferred onto each sterile cereal media. Inoculated media were incubated in the dark at $30 \pm 2°C$.

Sample collection

The mycelium-covered cereal samples were collected in the day 0, 2, 4, 6 and 8, then freeze dried with a Christ freeze dryer model beta 2-8 LSC-plus (Martin Christ Gefriertrocknungsanlagen GmbH, Osterode, Germany) to obtain the dried samples with the moisture less than 10%. Dried samples were separately ground into fine powders by using a Cemotec 1,090 miller (Tecator, Hoganas, Sweden) before keeping in a desiccator with no light until further analyses.

Determination of mycelium biomass

The biomass of *S. commune* was indirectly measured by determining the glucosamine content. A conversion method of chitin in mycelium to glucosamine was modified from Tsuji et al. (1969). Briefly, 100 mg of ground sample was added with 5 mL of 2 M HCl and then boiled ($100 \pm 2^{\circ}$ C) for 2 h. The mixture was placed in room temperature for cooling down before adjusting the volume to 5 mL with distilled water. A 1 mL of the supernatant was subjected to a volumetric flask and added a drop of alcohol solution of 0.5% (w/v) phenolphthalein, then followed carefully added 3 M NaOH until the mixture color changed to pink. The back titration was performed with 1% (w/v) KHSO₄ until the pink color disappeared. The titrated mixture was then adjusted with distilled water to 5 mL of volume.

Glucosamine in sample was analyzed by a method of Swift (1973) with Ehrlich's reagent. A 1 mL of the digested sample was transferred into a glass test tube. Then, a 1 mL of acetylacetone reagent (1 mL of acetylacetone mixed with 50 mL of 0.5 M Na₂CO₃) was added and the mixture was brought to boil for 20 min. After cooling, the mixture was added with 6 mL of ethanol and 1 mL of Ehrlich's reagent (2.67 g of *p*-dimethylaminobenzaldehyde dissolved in 15 mL of ethanol and 15 mL of Conc. HCl). The solution was incubated in water bath at 65°C for 10 min and then cooled before measuring the absorbance at 530 nm with a spectrophotometer (BioTek, Vermont, U.S.).

Sample extraction

Aqueous extraction of sample was carried out. Briefly, a 3 g of sample was added with 30 mL of distilled water, then performed the extraction in a shaking water bath at 42.5°C with a shaking speed of 160 rpm for 195 min. Then, the suspension was centrifuged at 3,857 g for 10 min. The supernatant was collected and filtered through a Whatman No.1 filter paper (Camlab, UK), then kept in an amber glass bottle at 4 ± 2 °C until use.

Determination of total phenolic content

Total phenolic content (TPC) was determined according to an assay of Kittisakulnam et al. (2016). A 25 μ L of obtained supernatant was mixed with 125 μ L of 0.2 M Folin-Ciocalteu reagent in a 96-well microplate. After leaving at room temperature (25 ± 2°C) for 5 min, the mixture was added with 100 μ L of 20% (w/v) Na₂CO₃, and then was allowed to stand in the dark at room temperature for 30 min. The absorbance

at 760 nm was measured by using a Varian Cary 50 MPR microplate reader (BioTek, Winooski, VT, USA). The total phenolic contents in all samples were expressed as mg gallic acid equivalents (GAE) per g dried extract.

Determination of total flavonoid content

Total flavonoid content (TFC) in all samples was analyzed by a modified method of Shoib et al. (2015). Briefly, a 150 μ L of the supernatant was placed in a 96-well microplate and mixed with 5 μ L of 5% (w/v) NaNO₂. After 5 min later, a 10 μ L of 10% (w/v) AlCl₃ was subsequently added into the mixture. After standing for another 6 min, a 70 μ L of 1 M NaOH was added and mixed well before measuring the absorbance at 510 nm. The total flavonoid content was expressed as mg catechin equivalents per g dried extract.

Antioxidant activity analyses

DPPH radical scavenging was analyzed by using an adapted method of Chen et al. (2020). Each sample supernatant was mixed with 100 μ M ethanolic DPPH solution in a ratio of 1:1 in a 96-well microplate before incubating at 37°C for 30 min without light. The incubated solution was measured an absorbance at 517 nm. The result was showed as an inhibitory concentration giving 50% of radical inhibition (IC₅₀).

ABTS cation radical scavenging was performed followed an assay of Kleekayai (2014). The ABTS stock solution was prepared by mixing 7 mM aqueous ABTS solution and 2.45 mM potassium persulfate at a ratio of 2:1, and then keeping in the dark for at least 16 h before use. The ABTS•⁺ solution was diluted with 5 mM phosphate buffer saline, pH 7.4 to obtain a 734 nm absorbance of 0.70 ± 0.02. A 10 μ L of sample supernatant was mixed with 200 μ L diluted ABTS•⁺ solution and followed incubated without light at room temperature for 6 min before measuring the absorbance at 734 nm. The result was shown as the IC₅₀.

Ferric reducing antioxidant power (FRAP) was analyzed by a modified method of Kittisakulnam et al. (2016). A 26 μ L of sample supernatant was mixed with 87 μ L of 0.2 M phosphate buffer, pH 6.6 and 87 μ L of 1% (w/v) potassium ferricyanide (in 0.2 M phosphate buffer, pH 6.6) and then incubated at 50°C for 20 min. An absorbance was measured to monitor the Perl's Prussian blue formation at 700 nm. The result of FRAP was reported as an efficient concentration giving 50% of total reducing power (EC₅₀).

Statistical analysis

One-way analysis of variance (ANOVA) in SPSS 18 software (SPSS Inc., Chicago, IL, USA) was used to analyze the data with Duncan's multiple range test (P < 0.05) which was used to determine the significant differences between variables.

RESULTS

Mycelium biomass

Determination of mycelium biomass in the solid-state fermentation was performed by indirectly measuring the glucosamine content which indicated the growth of mushroom fungi. The results of glucosamine content during the growth of *S. commune* in day 0-8 onto different cereal media are given in Table 1. The fungal growth increased along with the fermentation day. The highest glucosamine content on day 8 was 7.70 \pm 0.43 μ g/g DW and 7.25 \pm 0.12 μ g/g DW in jasmine rice and barley media, respectively. There is no significant difference between day 6 and day 8 of fungal biomass in barley. The lowest glucosamine contents were observed in sorghum and yellow dent corn which gave the mycelium biomass in a range of 1.52 - 1.76 μ g/g DW at day 8.

Total phenolic and flavonoid contents

The result in Table 1 shows that TPC of *S. commune* cultured on cereal media continuously increased along the growth period and the highest amounts were obtained on day 8. *S. commune* on wheat significantly exhibited the highest TPC production with

8.56 ± 1.09 mg GAE/g DW at days 8. The production of TFC was also correlated to the growing time, except in riceberry rice and sorghum which decreased in an early stage of fermentation. The highest quantity of TFC was found in barley with 577.35 ± 29.93 μ g CE/g DM, while the lowest TPC and TFC were presented in jasmine rice with 2.90 ± 0.44 mg GAE/g DW and 64.33 ± 3.17 μ g CE/g DW, respectively.

Antioxidant activity

The antioxidant activities of different cereal media cultured with *S. commune* were presented in Table 1. The results were reported in terms of IC₅₀ and EC₅₀ which the low value indicating to high antioxidant ability, and in contrast to high value, shows low activity. Interestingly, antioxidant properties of all fungal fermentations in 3 assays (DPPH, ABTS and FRAP) sharply increased within 2-4 d. Cultured wheat medium on day 8, which contained the highest TPC exhibited the highest antioxidant activities against DPPH• and ABTS•⁺ with the IC₅₀ of 3.83 ± 0.33 mg/mL and 2.38 ± 0.45 mg/mL, respectively. At day 8, the lowest DPPH• and ABTS•⁺ scavenging ability were observed in jasmine rice medium, which had the lowest TPC.

Table 1. Changes of fungal biomass, TPC, TFC and antioxidant activities of mycelial *S. commune* on different cereal media

| $\begin{array}{c c c c c c c c c c c c c c c c c c c $ | Cereal | Glucosamine | ТРС | TFC | IC ₅₀ and EC ₅₀ (mg/mL) | | |
|--|-------------|-----------------------------|-----------------------------|--|---|--|--|
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | media | content(µg/g) | (mg GAE/g) | (µg CE/g) | DPPH | ABTS | FRAP |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | Day 0 | | • • • • | | | | |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | SG | - | 0.82 ± 0.04^{b} | 317.02 ± 29.25 ^b | 27.71 ± 0.44 ^c | 81.19 ± 3.75 ^b | 63.91 ± 0.40° |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | YC | - | 0.90 ± 0.04^{ab} | 158.73 ± 11.57° | $57.74 \pm 1.90^{\circ}$ | 71.08 ± 2.77^{b} | 41.96 ± 2.91^{b} |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | BL | - | 0.22 ± 0.03^{d} | 95.39 ± 15.38d ^e | 80.32 ± 3.11^{f} | $180.80 \pm 7.06^{\circ}$ | 84.98 ± 4.12^{de} |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | 0A | - | $0.60 \pm 0.05^{\circ}$ | 127.39 ± 12.67^{cd} | 24.30 ± 1.52^{bc} | 62.93 ± 3.26^{ab} | 77.14 ± 2.37^{d} |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | WH | _ | $0.56 \pm 0.03^{\circ}$ | $138 19 \pm 3.92^{cd}$ | 38.70 ± 1.02^{d} | 61.83 ± 0.53^{ab} | 81.15 ± 1.01^{de} |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | 1R | _ | 0.12 ± 0.04^{d} | $53.06 \pm 2.61^{\circ}$ | 17244 + 6359 | $249\ 80\ +\ 21\ 45^{d}$ | 88 43 + 10 41° |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | MR | _ | 0.12 = 0.01 | $134\ 44\ +\ 3\ 00d^{\circ}$ | $1727 + 116^{b}$ | $83.08 \pm 2.46^{\circ}$ | $60.55 \pm 1.24^{\circ}$ |
| Name 1.01 ± 0.21 502.05 ± 22.11 5.27 ± 0.10^{-1} 49.71 ± 0.02^{-1} 10.55 ± 0.04^{-1} Day 2SG 0.35 ± 0.08^{bc} 0.97 ± 0.07^{a} 299.34 ± 13.27^{b} 25.20 ± 0.57^{c} 63.11 ± 0.52^{d} 54.80 ± 0.81^{c} YC 0.09 ± 0.14^{cd} 1.11 ± 0.10^{a} 149.89 ± 6.24^{c} 25.69 ± 3.14^{c} 51.07 ± 2.62^{c} 40.54 ± 0.20^{b} BL 1.70 ± 0.29^{a} 0.47 ± 0.13^{b} 131.18 ± 34.06^{c} 59.39 ± 0.35^{e} 51.11 ± 2.41^{bc} 61.66 ± 2.77^{de} OA 0.33 ± 0.02^{bc} 0.65 ± 0.03^{b} 153.88 ± 15.75^{c} 22.22 ± 1.86^{bc} 48.85 ± 0.22^{bc} 56.44 ± 2.60^{d} WH 0.51 ± 0.22^{b} 1.05 ± 0.25^{a} 146.05 ± 3.46^{c} 31.63 ± 1.36^{d} 46.11 ± 0.61^{b} 79.00 ± 1.25^{de} JR 0.04 ± 0.03^{d} 0.19 ± 0.03^{c} 55.49 ± 4.50^{d} 134.23 ± 1.33^{f} 98.77 ± 0.35^{e} 66.57 ± 6.01^{e} | RR | _ | $1 01 + 0 21^{a}$ | $902.99 + 22.11^{a}$ | $3.27 \pm 0.16^{\circ}$ | $49.71 \pm 0.02^{\circ}$ | $16.95 \pm 0.64^{\circ}$ |
| $ \begin{array}{cccccccccccccccccccccccccccccccccccc$ | Day 2 | | 1.01 - 0.21 | 502.55 - 22.11 | 5.27 ± 0.10 | 49.71 ± 0.02 | 10.95 ± 0.04 |
| SG 0.33 ± 0.02^{bc} 0.37 ± 0.07^{a} 233.54 ± 13.27^{c} 23.26 ± 0.37^{c} 03.11 ± 0.32^{c} 54.06 ± 0.31^{c} YC 0.09 ± 0.14^{cd} 1.11 ± 0.10^{a} 149.89 ± 6.24^{c} 25.69 ± 3.14^{c} 51.07 ± 2.62^{c} 40.54 ± 0.20^{b} BL 1.70 ± 0.29^{a} 0.47 ± 0.13^{b} 131.18 ± 34.06^{c} 59.39 ± 0.35^{e} 51.11 ± 2.41^{bc} 61.66 ± 2.77^{de} OA 0.33 ± 0.02^{bc} 0.65 ± 0.03^{b} 153.88 ± 15.75^{c} 22.22 ± 1.86^{bc} 48.85 ± 0.22^{bc} 56.44 ± 2.60^{d} WH 0.51 ± 0.22^{b} 1.05 ± 0.25^{a} 146.05 ± 3.46^{c} 31.63 ± 1.36^{d} 46.11 ± 0.61^{b} 79.00 ± 1.25^{de} JR 0.04 ± 0.03^{d} 0.19 ± 0.03^{c} 55.49 ± 4.50^{d} 134.23 ± 1.33^{f} 98.77 ± 0.35^{e} 66.57 ± 6.01^{e} | SG | 0 35 + 0 08 ^{bc} | 0 97 + 0 07ª | 299 34 + 13 27 ^b | 25 20 + 0 57° | 63 11 + 0 52 ^d | 54 80 + 0 81° |
| HereH | VC | 0.09 ± 0.00 | $1 11 + 0 10^{a}$ | $149.89 \pm 6.24^{\circ}$ | 25.20 ± 0.57 $25.69 \pm 3.14^{\circ}$ | $51.07 \pm 2.62^{\circ}$ | 40.54 ± 0.01 |
| DL 1.76 ± 0.29^{bc} 0.47 ± 0.13^{bc} 151.16 ± 54.06^{bc} 39.59 ± 0.33^{bc} 51.11 ± 2.41^{cc} 01.06 ± 2.77^{cc} OA 0.33 ± 0.02^{bc} 0.65 ± 0.03^{b} 153.88 ± 15.75^{cc} 22.22 ± 1.86^{bc} 48.85 ± 0.22^{bc} 56.44 ± 2.60^{d} WH 0.51 ± 0.22^{b} 1.05 ± 0.25^{a} 146.05 ± 3.46^{cc} 31.63 ± 1.36^{d} 46.11 ± 0.61^{b} 79.00 ± 1.25^{de} JR 0.04 ± 0.03^{d} 0.19 ± 0.03^{c} 55.49 ± 4.50^{d} 134.23 ± 1.33^{f} 98.77 ± 0.35^{e} 66.57 ± 6.01^{e} | RI | 1.70 ± 0.14 | 0.47 ± 0.10^{b} | $131 18 \pm 34 06^{\circ}$ | 50 30 ± 0.14 | 51.07 ± 2.02 $51.11 \pm 2.11^{\text{bc}}$ | $61.66 \pm 2.77^{\text{de}}$ |
| WH 0.51 ± 0.22^{b} 1.05 ± 0.25^{a} 146.05 ± 3.46^{c} 31.63 ± 1.36^{d} 46.11 ± 0.61^{b} 79.00 ± 1.25^{de} JR 0.04 ± 0.03^{d} 0.19 ± 0.03^{c} 55.49 ± 4.50^{d} 134.23 ± 1.33^{f} 98.77 ± 0.35^{e} 66.57 ± 6.01^{e} | | 0.33 ± 0.02 | 0.47 ± 0.13 | 151.10 ± 54.00 $153.00 \pm 15.75^{\circ}$ | 33.33 ± 0.33 | $12 85 \pm 0.22$ | 56.44 ± 2.00 |
| With 0.31 ± 0.22 1.05 ± 0.25 140.05 ± 3.46 31.05 ± 1.36 40.11 ± 0.01 79.00 ± 1.25 JR 0.04 ± 0.03^d 0.19 ± 0.03^c 55.49 ± 4.50^d 134.23 ± 1.33^f 98.77 ± 0.35^e 66.57 ± 6.01^e | | 0.55 ± 0.02 | 0.05 ± 0.05 | 135.00 ± 15.75 | 22.22 ± 1.00 | 40.00 ± 0.22 | 30.44 ± 2.00 |
| $JR = 0.04 \pm 0.05^{\circ} = 0.19 \pm 0.05^{\circ} = 55.49 \pm 4.50^{\circ} = 154.25 \pm 1.55^{\circ} = 96.77 \pm 0.55^{\circ} = 60.57 \pm 0.01^{\circ}$ | | $0.31 \pm 0.22^{\circ}$ | 1.03 ± 0.23 | $140.05 \pm 3.40^{\circ}$ | $31.03 \pm 1.30^{\circ}$ | $40.11 \pm 0.01^{\circ}$ | $79.00 \pm 1.23^{\circ}$ |
| | | $0.04 \pm 0.03^{\circ}$ | $0.19 \pm 0.03^{\circ}$ | 55.49 ± 4.50^{-1} | 134.23 ± 1.33 | $90.77 \pm 0.35^{\circ}$ | $00.37 \pm 0.01^{\circ}$ |
| MR 0.10 ± 0.04^{-60} $0.45 \pm 0.03^{\circ}$ $132.78 \pm 3.90^{\circ}$ $19.10 \pm 2.50^{\circ}$ $03.92 \pm 0.01^{\circ}$ $48.41 \pm 0.51^{\circ}$ | | 0.16 ± 0.04^{cd} | $0.45 \pm 0.03^{\circ}$ | $132.78 \pm 3.90^{\circ}$ | $19.10 \pm 2.50^{\circ}$ | $03.92 \pm 0.01^{\circ}$ | $48.41 \pm 0.51^{\circ}$ |
| RK 0.14 ± 0.05^{-5} 1.16 ± 0.02^{-5} 591.83 ± 12.45^{-5} 7.79 ± 0.93^{-5} 32.24 ± 0.28^{-5} 21.26 ± 1.65^{-5} | RK David | 0.14 ± 0.05^{co} | $1.16 \pm 0.02^{\circ}$ | $591.83 \pm 12.45^{\circ}$ | $7.79 \pm 0.93^{\circ}$ | $32.24 \pm 0.28^{\circ}$ | $21.26 \pm 1.65^{\circ}$ |
| | Day 4 | 0 40 L 0 70d | 1 70 1 0 050 | 241 10 1 10 10 ^{ab} | 21 42 4 1 2 4b | 11 04 J 1 00d | 20 1 4 L 2 4 5 d |
| SG $0.40 \pm 0.70^{\circ}$ $1.70 \pm 0.25^{\circ}$ $241.18 \pm 19.10^{\circ}$ $21.43 \pm 1.24^{\circ}$ $11.84 \pm 1.08^{\circ}$ $30.14 \pm 2.45^{\circ}$ | SG | $0.40 \pm 0.70^{\circ}$ | $1.70 \pm 0.25^{\circ}$ | 241.18 ± 19.10^{ab} | $21.43 \pm 1.24^{\circ}$ | $11.84 \pm 1.08^{\circ}$ | $30.14 \pm 2.45^{\circ}$ |
| YC 0.77 ± 0.34^{cd} 2.18 $\pm 0.59^{cd}$ 163.67 $\pm 6.77^{c}$ 15.62 $\pm 2.64^{c}$ 12.01 $\pm 1.42^{ce}$ 20.84 $\pm 0.35^{c}$ | YC | 0.77 ± 0.34^{cu} | 2.18 ± 0.59^{cd} | $163.67 \pm 6.77^{\circ}$ | $15.62 \pm 2.64^{\circ}$ | 12.01 ± 1.42^{ue} | $20.84 \pm 0.35^{\circ}$ |
| BL $3.63 \pm 0.78^{\circ}$ $3.13 \pm 0.31^{\circ}$ $1/6.11 \pm 30.92^{\circ}$ $12.82 \pm 0.20^{\circ}$ $5.78 \pm 0.25^{\circ}$ $1/.55 \pm 4.04^{\circ}$ | BL | $3.63 \pm 0.78^{\circ}$ | 3.13 ± 0.31^{ab} | $1/6.11 \pm 30.92^{ab}$ | $12.82 \pm 0.20^{\circ}$ | $5.78 \pm 0.25^{\circ}$ | $17.55 \pm 4.04^{\text{bc}}$ |
| OA $1.85 \pm 0.04^{\circ}$ $2.76 \pm 0.26^{\circ \circ}$ $253.17 \pm 54.04^{\circ}$ $9.25 \pm 0.13^{\circ}$ $6.73 \pm 0.04^{\circ}$ $11.10 \pm 1.56^{\circ \circ}$ | OA | $1.85 \pm 0.04^{\circ}$ | $2.76 \pm 0.26^{\text{bc}}$ | 253.1/ ± 54.04ª | $9.25 \pm 0.13^{\circ}$ | $6.73 \pm 0.04^{\circ}$ | 11.10 ± 1.56^{ab} |
| WH $1.45 \pm 0.25^{\circ c}$ 3.49 ± 0.57^{a} $190.84 \pm 11.61^{a \circ}$ 8.14 ± 0.14^{a} $7.38 \pm 0.89^{\circ c}$ 18.45 ± 1.79^{c} | WH | $1.45 \pm 0.25^{\text{DC}}$ | 3.49 ±0.57 ^a | 190.84 ± 11.61^{ab} | 8.14 ± 0.14^{a} | $7.38 \pm 0.89^{\text{pc}}$ | $18.45 \pm 1.79^{\circ}$ |
| JR $1.57 \pm 0.52^{\circ c}$ $1.79 \pm 0.27^{\circ e}$ 58.48 ± 0.72^{c} $22.35 \pm 0.12^{\circ}$ $13.69 \pm 0.36^{\circ}$ $27.16 \pm 2.68^{\circ}$ | JR | 1.57 ± 0.52 [∞] | $1.79 \pm 0.27^{\text{de}}$ | $58.48 \pm 0.72^{\circ}$ | 22.35 ± 0.12 ^₀ | $13.69 \pm 0.36^{\circ}$ | $27.16 \pm 2.68^{\circ}$ |
| MR $1.72 \pm 0.38^{\circ}$ $2.15 \pm 0.13^{\circ}$ $132.98 \pm 9.91^{\circ}$ $9.74 \pm 0.03^{\circ}$ $8.18 \pm 0.81^{\circ}$ $19.78 \pm 3.48^{\circ}$ | MR | 1.72 ± 0.38 ^b | 2.15 ± 0.13^{cd} | $132.98 \pm 9.91^{\circ}$ | $9.74 \pm 0.03^{\circ}$ | $8.18 \pm 0.81^{\text{DC}}$ | $19.78 \pm 3.48^{\circ}$ |
| RR $1.43 \pm 0.15^{\text{bc}}$ $2.44 \pm 0.11^{\text{cc}}$ $192.07 \pm 6.32^{\text{ab}}$ $12.45 \pm 0.74^{\text{a}}$ $9.22 \pm 0.15^{\text{c}}$ $9.34 \pm 2.57^{\text{a}}$ | RR | $1.43 \pm 0.15^{\text{DC}}$ | 2.44 ± 0.11^{cd} | 192.07 ± 6.32^{ab} | $12.45 \pm 0.74^{\circ}$ | 9.22 ± 0.15° | 9.34 ± 2.57^{a} |
| Day 6 | Day 6 | | | | | | |
| SG $1.52 \pm 0.08^{\circ}$ $3.43 \pm 0.13^{\circ}$ $191.79 \pm 16.85^{\circ}$ 7.36 ± 1.05^{ab} $7.71 \pm 1.14^{\circ}$ $9.26 \pm 0.29^{\circ}$ | SG | $1.52 \pm 0.08^{\circ}$ | 3.43 ± 0.13^{d} | 191.79 ± 16.85° | 7.36 ± 1.05^{ab} | $7.71 \pm 1.14^{\circ}$ | 9.26 ± 0.29° |
| YC $1.01 \pm 0.30^{\text{e}}$ $3.23 \pm 0.04^{\text{d}}$ $195.83 \pm 5.72^{\text{c}}$ $7.98 \pm 0.89^{\text{b}}$ $7.31 \pm 0.61^{\text{c}}$ $11.91 \pm 0.69^{\text{d}}$ | YC | 1.01 ± 0.30^{e} | 3.23 ± 0.04^{d} | 195.83 ± 5.72° | $7.98 \pm 0.89^{\circ}$ | $7.31 \pm 0.61^{\circ}$ | 11.91 ± 0.69^{d} |
| BL 6.89 ± 0.20^{a} 5.78 ± 0.70^{b} 555.15 ± 15.04^{a} 6.05 ± 0.41^{a} 4.32 ± 0.14^{b} 6.71 ± 0.94^{b} | BL | 6.89 ± 0.20ª | 5.78 ± 0.70^{b} | 555.15 ± 15.04ª | 6.05 ± 0.41^{a} | 4.32 ± 0.14^{b} | 6.71 ± 0.94 ^b |
| OA 3.72 ± 0.47^{d} 6.52 ± 1.16^{ab} 411.49 ± 59.23^{b} 5.80 ± 0.60^{a} 4.42 ± 0.07^{b} 4.64 ± 0.87^{a} | OA | 3.72 ± 0.47^{d} | 6.52 ± 1.16^{ab} | 411.49 ± 59.23 ^b | $5.80 \pm 0.60^{\circ}$ | 4.42 ± 0.07 ^b | 4.64 ± 0.87^{a} |
| WH $4.75 \pm 0.50^{\circ}$ 7.06 ± 0.38^{a} $255.09 \pm 17.07^{\circ}$ 6.12 ± 0.59^{a} 2.60 ± 0.85^{a} 6.69 ± 0.49^{b} | WH | 4.75 ± 0.50° | 7.06 ± 0.38^{a} | 255.09 ± 17.07° | 6.12 ± 0.59^{a} | 2.60 ± 0.85^{a} | 6.69 ± 0.49 ^b |
| JR $6.01 \pm 0.29^{\text{b}}$ $2.56 \pm 0.26^{\text{d}}$ $68.25 \pm 10.01^{\text{d}}$ $12.29 \pm 0.73^{\text{c}}$ $8.14 \pm 1.48^{\text{d}}$ $3.70 \pm 0.11^{\text{a}}$ | JR | 6.01 ± 0.29^{b} | 2.56 ± 0.26^{d} | 68.25 ± 10.01^{d} | 12.29 ± 0.73 ^c | 8.14 ± 1.48^{d} | 3.70 ± 0.11^{a} |
| MR $3.77 \pm 0.30^{\text{b}}$ $3.35 \pm 0.09^{\text{d}}$ $155.48 \pm 18.82^{\text{d}}$ $7.19 \pm 0.50^{\text{ab}}$ $8.61 \pm 0.24^{\text{c}}$ $4.35 \pm 0.44^{\text{a}}$ | MR | 3.77 ± 0.30^{b} | 3.35 ± 0.09^{d} | 155.48 ± 18.82^{d} | 7.19 ± 0.50^{ab} | 8.61 ± 0.24 ^c | $4.35 \pm 0.44^{\circ}$ |
| RR 3.37 ± 0.24^{d} 4.45 ± 0.27^{c} 212.36 ± 17.18^{c} 6.83 ± 0.10^{ab} 5.14 ± 0.90^{b} 3.73 ± 0.52^{a} | RR | 3.37 ± 0.24^{d} | 4.45 ± 0.27° | 212.36 ± 17.18° | 6.83 ± 0.10^{ab} | 5.14 ± 0.90^{b} | 3.73 ± 0.52ª |
| Day 8 | Day 8 | | | | | | |
| SG 1.76 \pm 0.14 ^d 4.08 \pm 0.16 ^{cd} 226.58 \pm 9.29 ^{cd} 5.21 \pm 0.51 ^a 4.73 \pm 0.92 ^c 7.17 \pm 0.35 ^d | SG | 1.76 ± 0.14^{d} | 4.08 ± 0.16^{cd} | 226.58 ± 9.29^{cd} | $5.21 \pm 0.51^{\circ}$ | 4.73 ± 0.92° | 7.17 ± 0.35^{d} |
| YC 1.52 ± 0.82^{d} 4.19 ± 0.35^{cd} 201.01 ± 11.31^{cd} 4.43 ± 0.92^{a} 7.05 ± 0.40^{d} 8.33 ± 0.48^{e} | YC | 1.52 ± 0.82^{d} | 4.19 ± 0.35^{cd} | 201.01 ± 11.31^{cd} | $4.43 \pm 0.92^{\circ}$ | 7.05 ± 0.40^{d} | $8.33 \pm 0.48^{\circ}$ |
| BI $7.25 \pm 0.12^{\circ}$ $6.87 \pm 1.28^{\circ}$ $577.35 \pm 29.93^{\circ}$ $4.53 \pm 0.27^{\circ}$ $3.71 \pm 0.72^{\circ}$ $2.14 \pm 0.23^{\circ}$ | BI | $7.25 \pm 0.12^{\circ}$ | 6.87 ± 1.28^{b} | $577.35 \pm 29.93^{\circ}$ | 4.53 ± 0.27^{a} | 3.71 ± 0.72^{bc} | 2.14 ± 0.23^{a} |
| $OA = 5, 39 \pm 0.38^{bc}$ $6, 10 \pm 0.51^{b}$ $515, 51 \pm 1.49^{b}$ $4, 45 \pm 0.37^{a}$ $3, 27 \pm 0.15^{ab}$ $3, 75 \pm 0.03^{ab}$ | 0A | 5.39 ± 0.38^{bc} | 6.10 ± 0.51^{b} | 51551 ± 149^{b} | $4 45 \pm 0.37^{\circ}$ | 3.27 ± 0.15^{ab} | 3.75 ± 0.03^{ab} |
| WH $6.00 \pm 0.20^{\text{b}}$ $8.56 \pm 1.09^{\text{a}}$ $290.81 \pm 7.11^{\text{c}}$ $3.83 \pm 0.33^{\text{a}}$ $2.38 \pm 0.45^{\text{a}}$ $4.66 \pm 0.21^{\text{c}}$ | WH | $6.00 \pm 0.20^{\circ}$ | $8.56 \pm 1.09^{\circ}$ | $290.81 \pm 7.11^{\circ}$ | $3.83 \pm 0.33^{\circ}$ | $2.27 \pm 0.45^{\circ}$ | $4.66 \pm 0.21^{\circ}$ |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | 1R | $7.70 \pm 0.43^{\circ}$ | 2.90 ± 0.44^{d} | $6433 \pm 317^{\circ}$ | $11.06 \pm 0.18^{\circ}$ | 7.54 ± 0.62^{d} | $350 \pm 0.78^{\circ}$ |
| $MR = 5.44 + 0.28^{bc} = 4.13 + 0.70^{cd} = 237.44 + 15.30^{de} = 5.80 + 1.56^{a} = 7.36 + 0.04^{d} = 2.97 + 0.21^{ab}$ | MR | 5.44 ± 0.28^{bc} | $4 13 \pm 0.70^{cd}$ | 237 44 + 15 30 ^{de} | $5 80 + 1 56^{\circ}$ | 736 ± 0.02^{d} | 2.00 = 0.70 2.97 + 0.21 ^{ab} |
| RR $4.45 \pm 0.71^{\circ}$ $4.86 \pm 0.08^{\circ}$ $244.98 \pm 23.05^{\circ}$ $5.74 \pm 0.15^{\circ}$ $4.46 \pm 0.29^{\circ}$ 3.06 ± 0.18^{ab} | RR | $4 45 \pm 0.71^{\circ}$ | $4.86 \pm 0.08^{\circ}$ | $244.98 \pm 23.05^{\circ}$ | 5.00 ± 1.00 5.74 ± 0.15 ^a | $446 \pm 0.29c$ | 3.06 ± 0.18^{ab} |

Note: Different letters in the same column of each day are significantly different at P < 0.05. SG, sorghum; YC, yellow dent corn; BL, barley; OA, oat; WH, wheat; JR, jasmine rice; MR, Mun-Poo rice; RR, riceberry rice

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FRAP assay could represent the ability of selected sample to reduce ferric iron (Fe³⁺) to ferrous iron (Fe²⁺). Reducing capacities of all cultured media also strongly increased along with the fermentation period. The highest FRAP activities with a EC₅₀ range of 2.14-3.75 mg/mL were obtained from barley, oat, Mun-Poo and riceberry rice media, while the lowest one found in yellow corn with an EC₅₀ of 8.33 ± 0.48 mg/mL.

DISCUSSIONS

Mycelium biomass

The difference of mycelium growth depended on the specific nutrient content in culture media. The most important factor is an adequate amount of carbohydrate and nitrogen sources in the media. Beloshapka et al. (2016) reported that jasmine rice and pearl barley consisted of high starch proportion up to 88% and 73% DM which could be digested into dextrin by glucoamylase of *S. commune* (Shimazaki et al., 1984). According to the reports of Ibrahim et al. (2011) suggested that dextrin is a good carbon source and positively promotes the growth of *S. commune* more than other carbohydrates such as monosaccharides, disaccharides, sugar alcohols and soluble starch.

In addition, Alam et al. (2010) found that organic nitrogen sources are more effective on the mycelium growth than inorganic nitrogen sources. This results were supported by a report of Hoa et al. (2015) suggested that mycelium density of *Pleurotus ostreatus* and *P. cystidiosus* cultured in the organic nitrogen containing medium was denser than that in the inorganic nitrogen containing medium. Several cereal grains are the sources of amino acids which considered as the organic nitrogen compounds (Cervantes-Pahm et al., 2014). Mosse (1990) reported that barley and oat consisted of high nitrogen content up to 4.01% and 3.80%, respectively which supported mycelium growth in this result. Nasreen et al. (2015) exhibited that the gram powder was a suitable nitrogen source for the growth of *S. commune.* Adebayo-Tayo et al. (2011) also reported that glycine was the most suitable nitrogen source for mycelium growth of mushroom *Macrolepiota procera* and *P. ostreatus*.

The digestion resistance of full-grain cereal considerably retards the fungal catabolism on conversion of the nutrients to the consumable energy. Jasmine rice and barley are normally passed the milling process which removes the hardly digestible polysaccharides containing parts including husk and bran. The starch digestion was rapidly occurred in barley medium causing *S. commune* mycelium sharply grew with significant difference among other cereals from day 2 until day 6. In contrast, the slowest growth was observed in corn which has a thick seed coat consisting of cellulose, xylose, and lignin (Knudsen, 2014). Similarly, Tripathy et al. (2009) observed that thickness of seed coat have effect to mycelium growth.

Interestingly, the phenolic compounds derived from cereal could slightly restrain the mycelium growth. High phenolic compounds containing cereal like riceberry rice, yellow dent corn and red sorghum significantly showed less fungal biomass than the low phenolic compounds containing cereals (jasmine rice and pearl barley). Normally, phenolic compounds are abundant in cereal bran fraction, therefore the cereals passed the bran removing procedure, contain lower in them. Upadhyay et al. (2016) reported that phenol addition in media inhibited the mycelial biomass production of basidiomycetous fungi. With an agreement of Yildiz et al. (2017), they found that there was an inverse relation between concentration of antioxidant compounds and growth rate of *P. ostreatus* and *P. citrinopileatus* (oyster mushroom).

Total phenolic and flavonoid contents

The increases of TPC and TFC were closely related to the extracellular enzyme activity that digested the culture materials and released the bound nutrients which consequently were used for antioxidant synthesis. Mostly, cell walls of cereal grains were digested with extracellular enzymes such as depolymerase, glycosidase, esterase, xylanase, laccase and peroxidases, and then released the free phenol compounds (Bonnin et al., 2002; McCUE et al., 2003). Whole grain wheat medium was rich in lignin which was in bran fraction, after lignin digestion by the fungus enzymes, the cinnamic acid derivatives were released resulting in the increment of TPC (Knudsen, 2014).

Moreover, *S. commune* could produce the antioxidant compounds by using shikimate pathway. This pathway generally was used for generating precursor chorismate from a central metabolic pathway, to eventually form antioxidative secondary metabolites, such as coumarin derivatives, isoflavones, flavones, flavonols, anthocyanins, tannins and other phenolic compounds. The carbon substrates particularly glucose could flow into this pathway via glycolysis. Barros et al. (2008) reported that glucose was the best carbon source for synthesizing phenol and flavonoid compounds in *Leucopaxillus giganteus* mycelium. Additionally, Tsujiyama (2009) also documented that the increase in phenolic compound production of *S. commune* was positively supported by increment of arabinoxylan content in solid state media. Arabinoxylan is mostly found in wheat, barley and oat (Knudsen, 2014).

The result was remarkably found that the reduction of TFC in riceberry rice and sorghum at the early period of the fungal growth. The most of flavonoid compounds in riceberry rice and red sorghum are anthocyanins belonging to flavonoids which characterized with heavy pigmented outer layer and they are attached to the cell wall sugars like arabinose, xylose, and glucose with the glycosidic bonds (Sivamaruthi et al., 2018). After cleaving these bonds by fungal β -glucosidases, the free anthocyanins are released and possibly oxidized by the extracellular phenol oxidase (laccase) and peroxidase of *S. commune* resulting in the decrease of TFC (Tovar-Herrera et al., 2018).

Antioxidant activity

DPPH, ABTS and FRAP have a positive correlation between antioxidant activity and TPC was observed in all cultured cereal media. There have been various studies establishing that the antioxidative activities were correlated to TPC. Mirfat et al. (2010) found that the fruiting body extracts of *S. commune* had the DPPH• scavenging activity with strong correlation to TPC. This also was in an agreement in other edible mushrooms (Cheung et al., 2003).

The results from our observation suggested that the antioxidant activities of *S. commune* mycelium onto edible cereals mostly came from the phenolic compounds. In addition, other bioactive compounds that formed during fermentation period beside phenolic and flavonoid compounds, such as schizophyllan (Li et al., 2011), amino acids (Anraku et al., 2015) and vitamins (Adejoye et al., 2007), were also reported to involve in antioxidant mechanisms.

CONCLUSIONS

The *S. commune* mycelium growth and its antioxidant properties were strongly affected by the nutrient compositions derived from each edible cereal. The starch-rich cereal sources like barley and jasmine rice were the most suitable media for *S. commune* mycelial growth. For the antioxidant production, wheat had the highest TPC which was corresponded with its highest DPPH• and ABTS•⁺ scavenging activities. The highest FRAP was obtained from barley which probable have other bioactive compounds from production of *S. commune* that show antioxidant activity. Therefore, solid-state fermentation of *S. commune* onto edible cereals could be effectively performed for antioxidative substance production. This finding has a potential to further develop for pharmaceutical and value-added foodstuffs in the future.

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